

YOUR CHEMISTRY LAB



SYLLABUS UPDATES FOR NEET-UG 2024

CHEMISTRY - CLASS-11th & 12th

CLASS-11 th				
Unit	Chapters	Status		
Unit 1	Some Basic Concepts of Chemistry	No Change		
Unit 2	Structure Of Atom	Change		
Unit 3	Classification of Elements and Periodicity	Change		
Unit 4	Chemical Bonding and Molecular Structure	No Change		
Unit 5	Thermodynamics	Change		
Unit 6	Equilibrium	No Change		
Unit 7	Redox Reactions	No Change		
Unit 8	The p -Block Elements	Change		
Unit 8	Organic Chemistry- Some Basic Principles and Techniques	No Change		
Unit 9	Hydrocarbons	No Change		
CLASS-12 th				
Unit	Chapters	Status		
Unit 11	Solutions	No Change		
Unit 12	Chemical Kinetics	No Change		
Unit 13	Electrochemistry	Change		
Unit 14	The p-Block Elements	Change		
Unit 15	The d-and f-Block Elements	Change		
Unit 16	Coordination Compounds	No Change		
Unit 17	Haloalkanes and Haloarenes	Change		
Unit 18	Alcohols, Phenols and Ethers	Change		
Unit 19	Aldehydes, Ketones and Carboxylic Acids	No Change		
Unit 20	Amines	Change		
Unit 21	Biomolecules	Change		







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CLASS-11 th					
Class-11 NCERT Chapter	Deleted Chemistry Topics/ Chapters	Added Chemistry Topics/Chapters			
Some Basic Concepts of Chemistry	_	-			
Structure of Atom	_	Elementary ideas of quantum mechanics, quantum mechanics, the quantum mechanical model of the atom, its important features. Concept of atomic orbitals as one-electron wave functions: Variation of Ψ and Ψ^2 with r for 1s and 2s orbitals			
Classification of Elements and					
Periodicity	_	_			
Chemical Bonding and Molecular					
Structure	_	_			
States Of Matter	Full Chapter Deleted	-			
Thermodynamics	Third Law of Thermodynamics				
	Enthalpy of Dilution	_			
Redox Reactions	_	-			
Hydrogen	Full Chapter Deleted	-			
The s-Block Elements	Full Chapter Deleted	-			
The p-Block Elements	Method of Preparation and Uses	_			
Organic Chemistry: Some Basic	_	_			
Principles and Techniques	_	_			
Hydrocarbons	_	-			
Environmental Chemistry	Full Chapter Deleted	_			







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CLASS-12 th					
Class-12 NCERT Chapter	Deleted Chemistry Topics/ Chapters	Added Chemistry Topics/Chapters			
The Solid State	Full Chapter Deleted	_			
Solutions	_	_			
Chemical Kinetics	_	-			
Electrochemistry	Corrosion	_			
Surface Chemistry	Full Chapter Deleted	_			
General Principles and Processes	Full Chapter Deleted				
of Isolation of Elements		_			
The p-Block Elements	_	-			
The d-and f-Block Elements	Chemical Reactivity of Lanthanoids	-			
Coordination Compounds	_	_			
Haloalkanes and Haloarenes	Environmental effects of dichloromethane,				
	tetrachloromethane	_			
Alcohols, Phenols and Ethers	Ether: Preparation Physical and Chemical	-			
	Properties				
Aldehydes, Ketones and	_	_			
Carboxylic Acids					
	Cyanides Isocyanides and Diazonium Salt:				
Amines	Preparation and Physical and Chemical	_			
	Properties				
Biomolecules	Polysaccharides (Starch Cellulose,	Biological Functions of Nucleic Acids			
2.1	Glycogen)	-			
Polymers	Full Chapter Deleted	_			
Chemistry in Everyday Life	Full Chapter Deleted	_			
Practical Chemistry	NEWLY ADDED:				
	1. Detection of extra elements (Nitrogen, Sulphur, Halogens) Inorganic Compounds;				
	Detection of the following functional group, hydroxyl (alcoholic and phenolic), carbonyl (aldehyde and ketones) carboxyl, and amino groups in organic compounds.				
	2. The chemistry involved in the preparation of the following: Inorganic compounds:				
	Mohr's salt. potash alum' Organic compounds: Acetanilide. p-nitro acetanilide, aniline				
	yellow, iodoform' The chemistry involved i	<u> </u>			
	and the use of indicators. oxalic-acid vs KMnO ₄ . Mohr's salt vs KMnO ₄ .				
	3. Chemical principles involved in the qualit				
	Cations - Pb ²⁺ .Cu ²⁺ , Al ³⁺ , Fe ³⁺ .Zn ²⁺ , Ni ²⁺ , Ca ²⁺ , Ba ²⁺ , Mg ²⁺ .NH ₄ ⁺				
	Anions - CO ₃ ²⁻ , S ²⁻ , SO ₄ ²⁻ , NO ₃ ⁻ NO ₂ ⁻ , CI ⁻ , Br ⁻ , I ⁻ (Insoluble salts excluded).				
	Chemical principles involved in the following experiments:				
	a) Enthalpy of solution of CuSO ₄				
	b) Enthalpy of neutralization of strong acid a				
	c) Preparation of lyophilic and lyophobic sols.				
	d) Kinetic study of the reaction of iodide	ions with hydrogen peroxide at room			
	temperature				



